



Material Safety Data Sheets

1. IDENTIFICATION

Product Name Oxalic acid, dihydrate

Other Names Ethanedioic acid, dihydrate; Oxalic acid, dihydrate

Uses No Data Available
Chemical Family No Data Available
Chemical Formula C2H2O4.2H2O

Chemical Name Ethanedioic acid, dihydrate

Product Description No Data Available
Company Arman sina.co
Contact Information info@armansina.co

info@armansina.com www.armansina.com

2. HAZARD IDENTIFICATION

Hazard Categories Acute Toxicity (Oral) - Category 4

Acute Toxicity (Dermal) - Category 4

Serious Eye Damage/Irritation - Category 1

Signal Word Danger

Hazard Statements H302 Harmful if swallowed.

H312 Harmful in contact with skin.
H318 Causes serious eye damage.

Precautionary Statements Prevention P270 Do not eat, drink or smoke when using this product.

P280 Wear protective gloves/protective clothing/eye protection/face protection.

Response P312 Call a POISON CENTER or doctor/physician if you feel unwell.

P330 Rinse mouth.

P302 + P352 IF ON SKIN: Wash with plenty of soap and water.

P363 Wash contaminated clothing before reuse.

P305 + P351 + P338 IF IN EYES: Rinse cautiously with water for several minutes. Remove contact

+ P310

lenses, if present and easy to do. Continue rinsing. Immediately call a POISON

CENTRE/doctor.

Disposal P501 Dispose of contents/container in accordance with local / regional / national /

international regulations.

Symbol







3. COMPOSITION/INFORMATION ON INGREDIENTS

Ingredients

Chemical Entity	Formula	CAS Number	Proportion
Oxalic acid, dihydrate	C2H2O4.2H2O	6153-56-6	<=100 %

4. FIRST AID MEASURES

Description of necessary measures according to routes of exposure

Swallowed IF SWALLOWED: Rinse mouth, then drink plenty of water. Do NOT induce vomiting. Call a Poison Centre or

doctor/physician for advice. Never give anything by mouth to an unconscious person.

IF IN EYES: Immediately flush eyes continuously with running water for several minutes, holding eyelids open and Eye

occasionally lifting the upper and lower lids. Immediately call a Poison Centre or doctor/physician for advice. Remove contact lenses if present and easy to do. Continue flushing until advised to stop by a Poison Information Centre or a

doctor, or for at least 15 minutes.

Skin IF ON SKIN (or hair): Remove contaminated clothing and shoes immediately. Flush skin and hair with running water

for at least 15 minutes; Wash with plenty of soap and water. Get medical advice/attention. Wash contaminated

clothing and shoes before reuse.

Inhaled IF INHALED: Remove victim to fresh air and keep at rest in a position comfortable for breathing. Get medical

advice/attention. Apply resuscitation if victim is not breathing - Administer oxygen if breathing is difficult.

Advice to Doctor Treat symptomatically. Ensure that attending medical personnel are aware of the identity and nature of the product(s)

involved, and take precautions to protect themselves.

Medical Conditions Aggravated No information available.

by Exposure

5. FIRE FIGHTING MEASURES

General Measures If safe to do so, move undamaged containers from fire area. Cool containers with water spray until well after fire is

Flammability Conditions Combustible material: May burn but does not ignite readily.

Extinguishing Media Use dry chemical, Carbon dioxide (CO2), foam or water spray for extinction. Use extinguishing measures that are

appropriate to local circumstances and the surrounding environment.

Fine dust dispersed in air in sufficient concentrations, and in the presence of an ignition source is a potential dust Fire and Explosion Hazard

explosion hazard.

Hazardous Products of

Combustion

Fire or heat will produce irritating, toxic and/or corrosive gases, including Carbon monoxide, Carbon dioxide, Formic

Special Fire Fighting

Instructions

Contain runoff from fire control or dilution water - Runoff may pollute waterways.

Personal Protective Equipment Wear self-contained breathing apparatus (SCBA) and chemical splash suit. SCBA and structural firefighter's uniform

may provide limited protection.

Flash Point No Data Available No Data Available **Lower Explosion Limit Upper Explosion Limit** No Data Available

No self-ignition below 400 ° C **Auto Ignition Temperature**

Hazchem Code No Data Available

6. ACCIDENTAL RELEASE MEASURES

Ensure adequate ventilation. ELIMINATE all ignition sources. Do not touch or walk through spilled material. Avoid **General Response Procedure**

generating dust. Avoid breathing dust and contact with eyes, skin and clothing.

Clean Up Procedures Collect material (sweep or vacuum up) and place into suitable containers for later disposal (see SECTION 13). Avoid

dispersal of dust in the air (i.e. clearing dusty surfaces with compressed air). Non-sparking tools should be used.

Containment Stop leak if safe to do so - Prevent entry into waterways, drains or confined areas. Prevent dust cloud.

Decontamination Wash away remainder with plenty of water. Prevent entry into drains and waterways. **Environmental Precautionary**

Measures

Evacuation Criteria Spill or leak area should be isolated immediately. Keep unauthorised personnel away.

Personal Precautionary

Measures

Use personal protective equipment as required (see SECTION 8).

7. HANDLING AND STORAGE

Handling Safety showers and eyewash facilities should be provided within the immediate work area for emergency use. Ensure

> adequate ventilation. Handle in accordance with good industrial hygiene and safety practice. Minimise dust generation and accumulation. Avoid breathing dust and contact with eyes, skin and clothing. Do not ingest. Use personal protective equipment as required (see SECTION 8). Dry powders can build static electricity charges when subjected to the friction of transfer and mixing operations. Provide adequate precautions, such as electrical

grounding and bonding, or inert atmospheres.

Store in a cool, dry and well-ventilated place, out of direct sunlight. Keep container tightly closed. Avoid exposure to Storage

air and moisture (hygroscopic). Keep away from heat and sources of ignition - No smoking. Keep away from

food/feedstuffs and incompatible materials (see SECTION 10).

Container Keep in the original container.

8. EXPOSURE CONTROLS / PERSONAL PROTECTION

General No specific exposure standards are available for this product. For Oxalic acid (CAS No. 144-62-7):

- Safe Work Australia (SWA) Exposure Standard: TWA = 1 mg/m3; STEL = 2 mg/m3.

- New Zealand Workplace Exposure Standard (WES): TWA = 1 mg/m3; STEL = 2 mg/m3.

- NIOSH REL: TWA = 1 mg/m3; ST = 2 mg/m3.

- OSHA PEL: TWA = 1 mg/m3.

- Immediately dangerous to life or health (IDLH) concentration: 500 mg/m3.

Exposure Limits No Data Available

Biological Limits No information available.

A system of local and/or general exhaust is recommended to keep employee exposures as low as possible. Local **Engineering Measures**

exhaust ventilation is generally preferred because it can control the emissions of the contaminant at its source,

preventing dispersion of it into the general work area.

Personal Protection Equipment - Respiratory protection: In case of inadequate ventilation, wear respiratory protection. Recommended: Organic

vapour/particulate (filter type A/P) respirator (refer to AS/NZS 1715 & 1716).

- Eye/face protection: Wear appropriate eye protection to prevent eye contact. Recommended: Face shield and safety glasses.

- Hand protection: Wear protective gloves. Recommended: Impervious gloves, e.g. Nitrile, neoprene, natural rubber, polyvinyl.

- Skin/body protection: Wear appropriate personal protective clothing to prevent skin contact. Recommended:

Standard work clothes, long pants, long sleeves, coveralls, safety shoes.

Special Hazards Precaustions No information available.

Work Hygienic Practices Do not eat, drink or smoke when using this product. Wash thoroughly after handling. Take off contaminated clothing

and wash before reuse

9. PHYSICAL AND CHEMICAL PROPERTIES

Solid **Physical State**

Appearance Crystals or powder

Odour Odourless

Colour Uncoloured or white

~0.7 (50 g/l) pН

Vapour Pressure 0.0312 Pa (@ 25 ° C) **Relative Vapour Density** No Data Available **Boiling Point** >160 ° C (Sublimes) **Melting Point** >160 ° C (Sublimes) No Data Available Freezing Point 108 g/L in water 25° C Solubility **Specific Gravity** 0.813 [EU A.3 method]

No Data Available Flash Point

No self-ignition below 400 ° C **Auto Ignition Temp**

Evaporation Rate No Data Available **Bulk Density** No Data Available **Corrosion Rate** No Data Available

>160 ° C **Decomposition Temperature**

No Data Available **Density Specific Heat** No Data Available No Data Available Molecular Weight **Net Propellant Weight** No Data Available No Data Available **Octanol Water Coefficient** No Data Available **Particle Size**

-1.7 (23 $^{\circ}$ C) [OECD Guideline 107] **Partition Coefficient**

Saturated Vapour Concentration No Data Available

Vapour Temperature No Data Available No Data Available Viscosity No Data Available **Volatile Percent VOC Volume** No Data Available **Additional Characteristics** No information available.

Potential for Dust Explosion Fine dust dispersed in air in sufficient concentrations, and in the presence of an ignition source is a potential dust

explosion hazard

Fast or Intensely Burning

Characteristics

No information available.

Flame Propagation or Burning

No information available.

Rate of Solid Materials

Non-Flammables That Could No information available. Contribute Unusual Hazards to a

Fire

Properties That May Initiate or Contribute to Fire Intensity

Combustible material; May burn but does not ignite readily.

Reactions That Release Gases

or Vapours

Fire or heat will produce irritating, toxic and/or corrosive gases, including Carbon monoxide, Carbon dioxide, Formic

acid

Release of Invisible Flammable No information available.

Vapours and Gases

10. STABILITY AND REACTIVITY

General Information The substance in solution is a medium-strong acid. Reacts violently with oxidants causing fire and explosion hazard.

Reacts with silver compounds, forming explosive silver oxalate. Attacks some forms of plastic.

Chemical Stability Stable under normal conditions of use and storage.

Conditions to Avoid Avoid generating dust. Avoid exposure to air and moisture. Keep away from heat and sources of ignition.

Materials to Avoid Incompatible/reactive with alkalis, alkaline solutions, ammonia, acid chlorides, halogenates, oxidising agents, metals.

Hazardous Decomposition

Products

Fire or heat will produce irritating, toxic and/or corrosive gases, including Carbon monoxide, Carbon dioxide, Formic

aid.

Hazardous Polymerisation Hazardous polymerisation will not occur.

11. TOXICOLOGICAL INFORMATION

- Acute toxicity: Harmful if swallowed and in contact with skin. Corrosive on ingestion; May cause effects on Calcium **General Information** balance. Signs of toxicity include nausea and vomiting, headaches, abdominal pain, diarrhoea, bloody stool,

numbness and tingling of fingers and toes, muscular irritability, tetany, convulsions, shock, cardiac irregularities and

circulatory collapse [NICNAS].

- Skin corrosion/irritation: Not irritating to skin. No skin irritation (Rabbit) [OECD TG 404].
- Eye damage/irritation: Causes serious eye damage. Irreversible effects on the eye (Rabbit) [OECD TG 405].
- Respiratory/skin sensitisation: Oxalic acid is not a skin sensitiser [OECD Guideline 429].
- Germ cell mutagenicity: Not considered to be genotoxic [NICNAS].
- Carcinogenicity: No evidence of carcinogenicity [NICNAS].
- Reproductive toxicity: Does not show specific reproductive or developmental toxicity [NICNAS].
- STOT (single exposure): Corrosion and irritant effects of the mouth and digestive tract, skin, eyes and respiratory tract have been reported following exposure to either the solid or concentrated solutions of oxalic acid [NICNAS].
- STOT (repeated exposure): May cause harmful cumulative effects (reduced thyroid function, renal toxicity, kidney damage/stone formation) following repeated oral exposure.
- Aspiration toxicity: No information available.

Acute

Ingestion Acute toxicity (Oral):

- LD50, Rat: >375 mg/kg bw. [Supplier's SDS].

Other Acute toxicity (Dermal):

- LD50, Rabbit: >20,000 mg/kg bw. [Supplier's SDS].

Carcinogen Category None

12. ECOLOGICAL INFORMATION

Ecotoxicity Aquatic toxicity:

- LC50, Fish (Leuciscus idus): 160 mg/l (96 h) [Supplier's SDS].

- EC50, Crustacea (Daphnia magna): 162.2 mg/l (48 h) [Supplier's SDS].

Persistence/Degradability Readily biodegradable.

Mobility No information available.

Environmental Fate Prevent entry into drains and waterways.

Bioaccumulation Potential No information available.

Environmental Impact No Data Available

13. DISPOSAL CONSIDERATIONS

General Information Dispose of contents/container via a licensed disposal company and in accordance with local/regional/national

regulations. Must not be disposed together with household garbage.

Special Precautions for Land Fill Contaminated packaging: Dispose of as unused product.

14. TRANSPORT INFORMATION

Land Transport

Proper Shipping Name
Class
No Data Available
Subsidiary Risk(s)
No Data Available
No Data Available
UN Number
No Data Available
Hazchem
No Data Available
Pack Group
No Data Available

Sea Transport

Special Provision

Proper Shipping Name Oxalic acid, dihydrate Class No Data Available Subsidiary Risk(s) No Data Available **UN Number** No Data Available Hazchem No Data Available **Pack Group** No Data Available **Special Provision** No Data Available No Data Available **EMS**

Marine Pollutant No

Comments NON-DANGEROUS GOODS: Not regulated for SEA transport.

No Data Available

Air Transport

Proper Shipping Name
Class
No Data Available
Subsidiary Risk(s)
No Data Available
UN Number
No Data Available
Hazchem
No Data Available
Pack Group
No Data Available
Special Provision
No Data Available

Comments NON-DANGEROUS GOODS: Not regulated for AIR transport.

15. OTHER INFORMATION

Revision 3

Reason for Issue

Update sds

Key/Legend

< Less Than > Greater Than

AICS Australian Inventory of Chemical Substances

atm Atmosphere

CAS Chemical Abstracts Service (Registry Number)

cm² Square Centimetres

CO2 Carbon Dioxide

COD Chemical Oxygen Demand deg C (° C) Degrees Celcius

deg F (° F) Degrees Farenheit

g Grams

g/cm³ Grams per Cubic Centimetre

g/I Grams per Litre

HSNO Hazardous Substance and New Organism IDLH Immediately Dangerous to Life and Health immiscible Liquids are insoluable in each other.

inHg Inch of Mercury inH2O Inch of Water

K Kelvin kg Kilogram

kg/m³ Kilograms per Cubic Metre

Ib Pound

LC50 LC stands for lethal concentration. LC50 is the concentration of a material in air which causes the death of 50% (one half) of a group of test animals. The material is inhaled over a set period of time, usually 1 or 4 hours. LD50 LD stands for Lethal Dose. LD50 is the amount of a material, given all at once, which causes the death of 50% (one half) of a group of test animals.

Itr or L Litre m³ Cubic Metre mbar Millibar mg Milligram

mg/24H Milligrams per 24 Hours mg/kg Milligrams per Kilogram mg/m³ Milligrams per Cubic Metre

Misc or Miscible Liquids form one homogeneous liquid phase regardless of the amount of either component

present.

mm Millimetre

mmH2O Millimetres of Water mPa.s Millipascals per Second

N/A Not Applicable

NIOSH National Institute for Occupational Safety and Health **NOHSC National Occupational Heath and Safety Commission**

OECD Organisation for Economic Co-operation and Development

Oz Ounce

PEL Permissible Exposure Limit

Pa Pascal

ppb Parts per Billion

ppm Parts per Million

ppm/2h Parts per Million per 2 Hours ppm/6h Parts per Million per 6 Hours

psi Pounds per Square Inch

R Rankine

RCP Reciprocal Calculation Procedure STEL Short Term Exposure Limit

TLV Threshold Limit Value

tne Tonne

TWA Time Weighted Average ug/24H Micrograms per 24 Hours

UN United Nations

wt Weight