



# **Material Safety Data Sheets**

# 1. IDENTIFICATION

Product Name Calcium chloride, dihydrate

Other Names No Data Available

Uses De-icing and freezing point depression; Road surfacing; Food additive; Laboratory and drying operations;

Miscellaneous applications.

Chemical Family No Data Available
Chemical Formula CaCl2.2H2O

Chemical Name Calcium chloride, dihydrate

Product Description No Data Available

Company Arman sina.co

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#### 2. HAZARD IDENTIFICATION

Hazard Categories Serious Eye Damage/Irritation - Category 2A

Signal Word Warning

Hazard Statements H319 Causes serious eye irritation.

Precautionary Statements Prevention P280 Wear eye protection/face protection.

P264 Wash face, hands and any exposed skin thoroughly after handling.

Response P337 + P313 If eye irritation persists: Get medical advice/attention.

P305 + P351 + P338 IF IN EYES: Rinse cautiously with water for several minutes. Remove contact

lenses, if present and easy to do. Continue rinsing.

Symbol



#### 3. COMPOSITION/INFORMATION ON INGREDIENTS

### Ingredients

Chemical Entity	Formula	CAS Number	Proportion
Calcium chloride, dihydrate	CaCl2.2H2O	10035-04-8	<=100 %

#### 4. FIRST AID MEASURES

Description of necessary measures according to routes of exposure

Swallowed IF SWALLOWED: Rinse mouth, then drink plenty of water. Do not induce vomiting. Call a Poison Centre or

doctor/physician for advice.

Eye IF IN EYES: Immediately flush eyes with running water for several minutes, holding eyelids open and occasionally

lifting the upper and lower lids. Remove contact lenses if present and easy to do. Continue rinsing for at least 15

minutes. If eye irritation persists, get medical advice/attention.

Skin IF ON SKIN: Wash with plenty of soap and water. Take off contaminated clothing and wash before reuse. If skin

irritation occurs, get medical advice/attention.

Inhaled IF INHALED: Remove victim to fresh air and keep at rest in a position comfortable for breathing until recovered. If

respiratory symptoms persist, get medical advice/attention.

Advice to Doctor Treat symptomatically.

Medical Conditions Aggravated

by Exposure

No information available.

#### 5. FIRE FIGHTING MEASURES

General Measures If safe to do so, move undamaged containers from fire area. Cool containers with water spray until well after fire is

out.

Flammability Conditions Non-combustible; Material does not burn.

Extinguishing Media If material is involved in a fire, use dry chemical, Carbon dioxide (CO2), foam or water spray for extinction. Use

extinguishing media appropriate for surrounding conditions.

Fire and Explosion Hazard Ambient fire may liberate hazardous vapours.

Hazardous Products of

Combustion

Fire or heat may produce irritating, toxic and/or corrosive fumes, including Calcium oxide, Hydrogen chloride.

Special Fire Fighting

Instructions

Personal Protective Equipment Wear self-contained breathing apparatus (SCBA) and chemical splash suit. SCBA and structural firefighter's uniform

Contain runoff from fire control or dilution water - Runoff may pollute waterways.

may provide limited protection.

Flash Point No Data Available
Lower Explosion Limit No Data Available
Upper Explosion Limit No Data Available
Auto Ignition Temperature No Data Available
Hazchem Code No Data Available

#### 6. ACCIDENTAL RELEASE MEASURES

Ensure adequate ventilation. Do not touch or walk through spilled material. Avoid generating dust. Avoid breathing General Response Procedure

dust and contact with eyes, skin and clothing.

Collect material (take up dry) and seal in properly labelled containers for disposal (see SECTION 13). Clean Up Procedures

\*Caution: May react exothermically on contact with water.

Containment Stop leak if safe to do so - Prevent entry into waterways, drains or confined areas.

Decontamination Wash area down with excess water.

**Environmental Precautionary** 

Measures

**Evacuation Criteria** Spill or leak area should be isolated immediately. Keep unauthorised personnel away.

Personal Precautionary

Measures

Use personal protective equipment as required (see SECTION 8).

#### 7. HANDLING AND STORAGE

Safety showers and eyewash facilities should be provided within the immediate work area for emergency use. Ensure Handling

> adequate ventilation. Handle in accordance with good industrial hygiene and safety practice. Minimise dust generation and accumulation. Avoid breathing dust and contact with eyes, skin and clothing. Do not ingest. Use

> Prevent entry into drains and waterways. If environmental contamination has occurred, advise local emergency

personal protective equipment as required (see SECTION 8).

Storage Store in a cool, dry and well-ventilated place, out of direct sunlight. Keep container tightly closed when not in use -

check regularly for spills. Protect from moisture (hygroscopic; reacts with water). Keep away from incompatible

materials (see SECTION 10).

Container Keep in the original, properly labelled container.

# 8. EXPOSURE CONTROLS / PERSONAL PROTECTION

General No specific exposure standards are available for this product. For dusts from solid substances without specific

occupational exposure standards:

- Safe Work Australia Exposure Standard (Nuisance dusts): 8 hr TWA = 10 mg/m3 (measured as inhalable dust). - New Zealand WES (Particulates not otherwise classified): TWA = 10 mg/m3; TWA = 3 mg/m3 (respirable dust).

**Exposure Limits** 

**Biological Limits** No information available. Engineering Measures A system of local and/or general exhaust is recommended to keep employee exposures as low as possible. Local

exhaust ventilation is generally preferred because it can control the emissions of the contaminant at its source,

preventing dispersion of it into the general work area.

Personal Protection Equipment - Respiratory protection: Wear respiratory protection in case of inadequate ventilation or an inhalation risk exists.

Recommended: Dust mask/particulate respirator (refer to AS/NZS 1715 & 1716).

- Eye/face protection: Wear appropriate eye protection to avoid eye contact. Recommended: Safety glasses with

side-shields, chemical goggles or full face-shield as appropriate (refer to AS/NZS 1337).

- Hand protection: Handle with gloves. Recommended: Impervious gloves, e.g. nitrile rubber (refer to AS/NZS 2161.1)

- Skin/body protection: Wear appropriate personal protective clothing to avoid skin contact. Recommended: Overalls

(cotton), safety shoes; Chemical-resistant apron when large quantities are handled.

Special Hazards Precaustions No information available.

Work Hygienic Practices Do not eat, drink or smoke when using this product. Wash face, hands and any exposed skin thoroughly after

handling. Wash contaminated clothing and other protective equipment before storage or re-use.

#### 9. PHYSICAL AND CHEMICAL PROPERTIES

Physical State Solid

Appearance Flakes, powder, granules, pellets

Odour Odourless
Colour White

pH 9 - 11 (5% CaCl2 soln. @ 20° C)

Vapour Pressure

No Data Available
Relative Vapour Density

No Data Available
Boiling Point

No Data Available
Melting Point

>=175 ° C

Freezing Point No Data Available

Soluble in water - Soluble in alcohols

Specific Gravity 1.85

Flash Point No Data Available
Auto Ignition Temp No Data Available
Evaporation Rate No Data Available
Bulk Density 800 - 900 kg/m3
Corrosion Rate No Data Available
Decomposition Temperature No Data Available

Density 1.85 g/cm3

Specific Heat No Data Available

Molecular Weight No Data Available

Net Propellant Weight No Data Available

Octanol Water Coefficient No Data Available

Particle Size No Data Available

Partition Coefficient No Data Available

Saturated Vapour Concentration No Data Available

Vapour Temperature No Data Available
Viscosity No Data Available
Volatile Percent No Data Available
VOC Volume No Data Available

Additional Characteristics Product is strongly hygroscopic.

Potential for Dust Explosion No information available.

Fast or Intensely Burning No information available.

Characteristics

No. 1 Company of the Company of the

Flame Propagation or Burning

Rate of Solid Materials

No information available.

Non-Flammables That Could Contribute Unusual Hazards to a

Properties That May Initiate or Contribute to Fire Intensity

Non-combustible; Material does not burn.

No information available.

No information available.

Reactions That Release Gases

or Vapours

Fire or heat may produce irritating, toxic and/or corrosive fumes, including Calcium oxide, Hydrogen chloride.

Release of Invisible Flammable

Vapours and Gases

# 10. STABILITY AND REACTIVITY

May react exothermically on contact with water. General Information Chemical Stability Stable under ordinary conditions of storage and use. Avoid dust formation. Protect from moisture/humidity. Conditions to Avoid

Materials to Avoid Incompatible/reactive with strong acids, strong bases, bromine trifluoride, water, zinc, polymerisable materials.

Hazardous Decomposition

**Products** 

Fire or heat may produce irritating, toxic and/or corrosive fumes, including Calcium oxide, Hydrogen chloride.

Hazardous Polymerisation Will not occur.

#### 11. TOXICOLOGICAL INFORMATION

General Information - Acute toxicity: May be harmful if swallowed; May cause gastrointestinal irritation, with abdominal pain, nausea,

vomiting and diarrhoea.

- Skin corrosion/irritation: Skin contact may cause irritation.

- Eye damage/irritation: Causes serious eye irritation; May cause redness, tearing, stinging, blurred vision. - Respiratory/skin sensitisation: Not expected to cause sensitisation of the respiratory tract or the skin.

- Germ cell mutagenicity: Not considered to be mutagenic.

- Carcinogenicity: Not considered to be carcinogenic.

- Reproductive toxicity: Not considered to be toxic to reproduction or development. - STOT (single exposure): Inhalation of dusts may cause respiratory tract irritation.

- STOT (repeated exposure): Repeated or prolonged exposure is not expected to cause specific target organ toxicity.

- Aspiration toxicity: Not expected to be an aspiration hazard.

Acute

Acute toxicity (Oral): Ingestion

For Calcium chloride (CAS No. 10043-52-4): - LD50, Rat (male): 2,120 - 3,798 mg/kg bw. - LD50, Rat (female): 2,361 - 4,179 mg/kg bw.

- LD50, Rat (combined male & female): 2,301 mg/kg bw. [ECHA].

Carcinogen Category None

#### 12. ECOLOGICAL INFORMATION

**Ecotoxicity** No information available. Persistence/Degradability No information available. Mobility No information available.

**Environmental Fate** Prevent entry into drains and waterways.

Bioaccumulation Potential No information available. Environmental Impact No Data Available

#### 13. DISPOSAL CONSIDERATIONS

General Information Dispose of contents/container via a licensed waste contractor and in accordance with local/regional/national

regulations.

Special Precautions for Land Fill No information available.

# 14. TRANSPORT INFORMATION

### Land Transport (New Zealand) NZS5433

Proper Shipping Name Calcium chloride, dihydrate

Class No Data Available
Subsidiary Risk(s) No Data Available
No Data Available

UN Number No Data Available
Hazchem No Data Available
Pack Group No Data Available
Special Provision No Data Available

Comments NON-DANGEROUS GOODS: Not regulated for LAND transport.

### Sea Transport IMDG Code

Proper Shipping Name Calcium chloride, dihydrate

Class No Data Available
Subsidiary Risk(s) No Data Available
UN Number No Data Available
Hazchem No Data Available
Pack Group No Data Available
Special Provision No Data Available
EMS No Data Available

Marine Pollutant No

Comments NON-DANGEROUS GOODS: Not regulated for SEA transport.

### Air Transport IATA DGR

Proper Shipping Name Calcium chloride, dihydrate

Class No Data Available
Subsidiary Risk(s) No Data Available
UN Number No Data Available
Hazchem No Data Available
Pack Group No Data Available
Special Provision No Data Available

Comments NON-DANGEROUS GOODS: Not regulated for AIR transport.

#### 15. OTHER INFORMATION

Revision 2

Key/Legend

< Less Than > Greater Than

atm Atmosphere

CAS Chemical Abstracts Service (Registry Number)

cm² Square Centimetres

CO2 Carbon Dioxide

COD Chemical Oxygen Demand deg C (° C) Degrees Celcius deg F (° F) Degrees Farenheit

g Grams

g/cm³ Grams per Cubic Centimetre

g/I Grams per Litre

HSNO Hazardous Substance and New Organism IDLH Immediately Dangerous to Life and Health immiscible Liquids are insoluable in each other.

inHg Inch of Mercury inH2O Inch of Water

K Kelvin kg Kilogram

kg/m³ Kilograms per Cubic Metre

lb Pound

LC50 LC stands for lethal concentration. LC50 is the concentration of a material in air which causes the death of 50% (one half) of a group of test animals. The material is inhaled over a set period of time, usually 1 or 4 hours. LD50 LD stands for Lethal Dose. LD50 is the amount of a material, given all at once, which causes the death of 50% (one half) of a group of test animals.

Itr or L Litre m³ Cubic Metre mbar Millibar mg Milligram

mg/24H Milligrams per 24 Hours mg/kg Milligrams per Kilogram

mg/m³ Milligrams per Cubic Metre

Misc or Miscible Liquids form one homogeneous liquid phase regardless of the amount of either component present.

mm Millimetre

mmH2O Millimetres of Water mPa.s Millipascals per Second

N/A Not Applicable

NIOSH National Institute for Occupational Safety and Health NOHSC National Occupational Heath and Safety Commission OECD Organisation for Economic Co-operation and Development

Oz Ounce

PEL Permissible Exposure Limit

Pa Pascal

ppb Parts per Billion ppm Parts per Million

ppm/2h Parts per Million per 2 Hours ppm/6h Parts per Million per 6 Hours

psi Pounds per Square Inch

R Rankine

RCP Reciprocal Calculation Procedure STEL Short Term Exposure Limit

TLV Threshold Limit Value

tne Tonne

TWA Time Weighted Average ug/24H Micrograms per 24 Hours

**UN United Nations** 

wt Weight